

Ans (a).

According to generalized version of
CFT \Rightarrow

$$\frac{P_F(+w)}{P_R(-w)} = e^{+w} \quad \text{--- ①}$$

w is entropy production of the
driven system measured over some time
interval.

$$w = -\beta \Delta F + \beta W \quad \text{--- ②}$$

Whenever eq. ① is valid \rightarrow

$$\langle e^{-w} \rangle = \int_{-\infty}^{+\infty} P_R(-w) dw = 1$$

So following relation holds \rightarrow

$$\langle e^{-w} \rangle = 1$$

Substitute the value of w from eq. ② \rightarrow

$$\langle e^{-(-\beta \Delta F + \beta W)} \rangle = 1$$

$$\langle e^{\beta \Delta F - \beta W} \rangle = 1$$

$$\because e^{a+b} = e^a \cdot e^b$$

$$\langle e^{\beta \Delta F} \cdot e^{-\beta W} \rangle = 1$$

Free energy difference is a state function,
and can be moved outside the average.

$$e^{\beta \Delta F} \langle e^{-\beta W} \rangle = 1$$

$$\langle e^{-\beta W} \rangle = 1/e^{\beta \Delta F}$$

$$\langle e^{-\beta W} \rangle = e^{-\beta \Delta F}$$

The above relation is known as
Jarzynski equality.

Ans (b).

- The vertical axis of the graph $P_U(W)$ denote the probability distribution of the values of the work performed on the molecule in an infinite number of pulling experiments along the unfolding (U) process, and define $P_R(W)$ analogously for the reverse (R) process.
- The unfolding and the refolding work distributions cross at a value of the work ΔG . The significance of the line is that the work done doesn't depend on pulling speed and the distributions also satisfy the CFT and they can be fitted to gaussian distributions.
- Irreversibility increases with the pulling speed and unfolding-refolding work distributions become progressively more separated. however, that the unfolding and the refolding distributions cross at a value of the work ΔG .

Ans ©.

- The folding energy for wild type is approximately $154.1 \pm 0.4 \text{ kBT}$ while for mutant type it is around $157.9 \pm 0.2 \text{ kBT}$.
So, the difference between these two is approximately $3.8 \pm 0.6 \text{ kBT}$.
- the slope of the inset curve in figure 3 (upper righthand corner, small graph) being 1.06 tells us that these molecules are ideal to test the validity of CFT equation in far-from equilibrium regime.
Following is the math behind it -

According to CFT \Rightarrow

$$\frac{P_U(w)}{P_R(-w)} = \exp\left(\frac{w - \Delta G}{k_B T}\right)$$

in given graph of $\log\left\{\frac{P_U(w)}{P_R(-w)}\right\}$ v/s $\frac{w}{k_B T}$

$$\text{slope} = 1.06 \approx 1$$

$$\log\left\{\frac{P_U(w)}{P_R(-w)}\right\} = 1 \times \frac{w}{k_B T} + C$$

$$\log\left\{\frac{P_U(w)}{P_R(-w)}\right\} = \frac{w}{k_B T} + C$$

$$\frac{P_U(w)}{P_R(-w)} = \exp\left(\frac{w}{k_B T} + C\right)$$

$$\frac{P_U(w)}{P_R(-w)} = \exp\left(\frac{w}{k_B T} + \left(\frac{-\Delta G}{k_B T}\right)\right)$$

Assuming

$$C = \frac{-\Delta G}{k_B T}$$

$$\boxed{\frac{P_U(w)}{P_R(-w)} = \exp\left(\frac{w - \Delta G}{k_B T}\right)}$$

So, overall idea is simple that slope is nearly equal to one which proves the CFT.